

**DETERMINATION OF CATECHOL IN NATURAL WATERS
WITH A BIOSENSOR BASED ON TYROSINASE IMMOBILIZED WITHIN
POLY-3,4-ETHYLENEDIOXYTHIOPHENE FILM**

Constantin Apetrei

*"Dunarea de Jos" University of Galati, Faculty of Science and Environment, Department of
Chemistry, Physics and Environment, 47 Domneasca Street, RO-800008, Galati, Romania*

Corresponding author: apetreic@ugal.ro

Abstract: Some phenolic compounds with low molecular mass are toxic pollutants. They have a potential hazard especially for aquatic life, they often enter the aquatic environments via industrial residues from different categories of production, such as dyes, drugs, plastics, resins, pesticides etc. The most important source of contamination is the paper and cellulose industry. Among phenolic compounds, the catechol is very important in the human health because it could react with protein and cause blood coagulation. Therefore, it is significant to improve detection of catechol to protect human health. The use of biosensors for the detection and quantification of catechol has advantages, such as fast response, short time of analysis, simplicity of fabrication, and high selectivity and sensitivity. Additionally, sample treatment is not required.

In this work, a biosensor using a screen printed carbon electrode (SPCE) modified with tyrosinase within a poly-3,4-ethylenedioxythiophene (PEDOT) film is developed. PEDOT layer was electrosynthesized in the presence of sodium sulphate as doping agent. Glutaraldehyde crosslinking agent was used as for tyrosinase immobilization. The developed biosensor was characterized by scanning electron microscopy (SEM), electrochemical impedance spectroscopy (EIS) and cyclic voltammetry (CV) in the presence of catechol. After optimization of the experimental parameters, the determination of catechol was carried out by amperometry at fixed potential. The analytical performances characteristics of the biosensor were determined obtaining a linear concentration range from 2.0×10^{-6} to $12.0 \times 10^{-5} \text{ mol} \cdot \text{L}^{-1}$ with a detection limit of $1.2 \times 10^{-7} \text{ mol} \cdot \text{L}^{-1}$. The proposed biosensor presented appropriate repeatability and stability for the practical applications. Furthermore, this new method based on amperometry was successfully used in the determination of catechol in Danube water samples. The results obtained with the biosensor were in agreement with a 99% confidence level for those achieved using the official spectrophotometric method.

Keywords: biosensor, catechol, natural water

Acknowledgement: This work was supported by the project "Strategy and actions for preparing the national participation in the DANUBIUS-RI Project" acronym "DANS" financed by the Romanian Ministry of Research and Innovation.